## **H2 PHYSICS DEFINITIONS LIST**

Term	Definition
	SECTION I: MEASUREMENT
	Chapter 1: Measurement
Scalar	A scalar quantity is one which has <u>magnitude</u> but <u>no</u> direction.
Vector	A vector is a quantity which has <u>direction as well as</u> magnitude.
S	CTION II: NEWTONIAN MECHANICS
	Chapter 2: Kinematics
Displacement, s	Total distance moved by an object along a particular direction.
Speed	The rate of change of distance with respect to time.
Velocity, v	The rate of change of displacement with respect to time.
Acceleration, a	The rate of change of velocity with respect to time.
Average speed/velocity	The <u>average rate</u> of change of distance/displacement with respect to time.
Instantaneous Velocity	The rate of change of displacement with respect to time at a particular time.
	Chapter 3: Dynamics
Newton's First Law	A body <u>continues in its state of rest or uniform motion in a</u> straight line unless a <u>resultant external force</u> acts on it.
Newton's Second Law	The <u>rate of change of momentum</u> of a body is <u>proportional</u> to the <u>resultant force</u> acting on it and occurs <u>in the direction</u> of the force.
Newton's Third Law	If body A exerts a force on body B, then body B exerts an equal but opposite force on body A.
Linear Momentum	The product of the mass of an object and its velocity.
Impulse	Impulse is the area under the force-time graph.
Force, F	The rate of change of momentum.
Principle of Conservation of Momentum	The total momentum of a system remains <u>constant</u> provided no external resultant forces act on the system.  OR  The total momentum of an <u>isolated system</u> of bodies is
Inertia	constant.  The reluctance of a body to start moving or to change its
	motion once it has started.
Equilibrium	When the <u>state</u> of an object remains <u>unchanged</u> even though two or more forces are acting on it.
Chapter 4: Forces	
Hooke's Law	The <u>force</u> needed to cause an extension/compression in a spring is <u>directly proportional to its extension/compression</u> .
Upthrust, U	It is the <u>upward force</u> acting on an object that is <u>partially or</u> fully immersed in a fluid. (RJCPromo07)
Equilibrium	A system is in equilibrium if there is no resultant force and

	no recultant torque acting on it
	no resultant torque acting on it.
Centre of Gravity	The point at which all the weight of an object may be
	considered to be acting as if the object were a particle.
Couple	A couple consists of a pair of parallel forces of equal
	magnitude but opposite direction whose lines of action do
	not coincide.
Moment of a Force	The moment of a force about a point is the product of the
	force with the perpendicular distance of the force from that
	point.
Torque of a Couple, τ	The torque of a couple is the product of one of the forces
	with the perpendicular separation between the couple.
Archimedes' Principle	An object immersed fully or partially in a fluid experiences a
*	buoyant force equal in magnitude to the weight of the fluid
	displaced.
Centre of Mass	The point at which all of the mass of an object or system
Centre of Mass	may be considered to be concentrated.
Principle of Flotation	An object floating in a fluid always displaces its own weight
· · · · · · · · · · · · · · · · · · ·	of fluid.
Rotational Equilibrium	A system is in rotational equilibrium if there is no resultant
notational Equilibrium	torque.
(	Chapter 5: Work, Energy and Power
Work Done	The product of a force and the displacement in the direction
Work Done	of the force.
Power	Work done per unit time.
	Chapter 6: Motion in a Circle
Angular Velocity, ω	The rate of change of angular displacement with respect to
, mganar variotity, a	time.
Centripetal Acceleration	Acceleration which is always perpendicular to the velocity
• • • • • • • • • • • • • • • • • • • •	and always acts towards the centre of the circular motion.
Uniform Circular Motion	The motion of an object moving in circular path at constant
	speed with constant angular velocity.
Centripetal Force	The resultant perpendicular force acting on an object
	moving in circular motion
	Chapter 7: Gravitational Field
Gravitational Field	A gravitational field due to a body is a region in space in
	which another body placed in the region experiences a force
	of attraction by the first body.
Newton's Law of	Newton's law of gravitation states that the force of
Gravitation/Gravitational	attraction between two point masses is directly
Force	proportional to the product of their masses and inversely
	proportional to the square of their distance apart.
Gravitational Field	The gravitational field strength at a point is the gravitational
Strength, g	force per unit mass experienced by a mass placed at that
S. C. B. J. B	point.
Gravitational Potential, φ	Gravitational potential at a point is the work done per unit
Gravitational Potential, ψ	mass by an external agent in bringing a mass from infinity to
	Times of an executar affert in punising a mass itom injusts to

	that point without a change in kinetic energy. (RJCPromo07)
Gravitational Potential	The Gravitational Potential Energy of a mass is defined as
Energy	the work done by an external agent in bringing the mass
	from infinity to its present location (without any change in
Complete Company of the Company of t	KE).
	Chapter 8: Oscillations
Amplitude	The maximum displacement from the equilibrium position.
Period, T	The time taken to complete one cycle of oscillation.
Frequency, f	The number of complete cycles per second made by the
	oscillating object.
Simple Harmonic Motion	The motion of the body whose acceleration is directly
	proportional to its displacement from a fixed point
	(equilibrium position) and is always directed towards that
	fixed point.
Resonance	The tendency of a system to oscillate at maximum
	amplitude at its natural frequency.
Forced Oscillation	When the system is forced to oscillate at a frequency other
	than the natural frequency by a periodic external force.
Natural Frequency	The frequency of oscillation when a system oscillates freely
	without any external force applied.
Displacement, s	The distance of the oscillating object from its equilibrium
Miles - Toxico veces and a second	position at any instant.
	SECTION III: THERMAL PHYSICS Chapter 9: Thermal Physics
Internal Energy	The sum of the microscopic kinetic and potential energies of
internal Lifergy	the molecules that make up the system.
Thermal Equilibrium	When two objects in thermal contact cease to have any
The man against an	exchange of heat.
Absolute Zero	The theoretical temperature at which the molecules of a
	substance have the lowest energy and hence, the substance
	has minimum internal energy.
Kelvin, K	The Kelvin is defined as 1/273.16 of the temperature
	difference between absolute zero and the triple point of
	water.
Specific Heat Capacity, c	It is the quantity of heat required to raise the temperature
specific fieur capacity, c	
	of 1kg of the body by 1K.
Latent Heat	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in
Latent Heat	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.
Latent Heat Specific Latent Heat of	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to
Latent Heat	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a
Latent Heat  Specific Latent Heat of Fusion	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a change in temperature.
Latent Heat  Specific Latent Heat of Fusion  Specific Latent Heat of	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a change in temperature.  It is the thermal energy required for 1kg of substance to
Latent Heat  Specific Latent Heat of Fusion	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a change in temperature.  It is the thermal energy required for 1kg of substance to change from the liquid phase to the gaseous phase without
Latent Heat  Specific Latent Heat of Fusion  Specific Latent Heat of Vaporisation	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a change in temperature.  It is the thermal energy required for 1kg of substance to change from the liquid phase to the gaseous phase without a change in temperature.
Latent Heat  Specific Latent Heat of Fusion  Specific Latent Heat of	of 1kg of the body by 1K.  It is the thermal energy required by matter for a change in phase.  It is the thermal energy required for 1kg of substance to change from the solid phase to the liquid phase without a change in temperature.  It is the thermal energy required for 1kg of substance to change from the liquid phase to the gaseous phase without

	Chapter 12: Electric Fields
	ECTION V: ELECTRICITY AND MAGNETISM
Coherence	Sources having constant phase difference.
	constructively and destructively respectively. (RJCPrelim07)
	to form regions of maxima (bright) and minima (dark) due to waves meeting
	region and time so as
	Interference is the superposition of waves in the same
	OR
	given by the <u>principle of superposition</u> . (RJCNotes)
	waves to give a resultant wave whose resultant amplitude is
Interference	Interference is the superposition of two or more coherent
	around an obstacle.
Diffraction	Diffraction is the bending of waves through an aperture of
	(RJCCT108)
ouper position.	displacement due to each wave arriving at that point
Superposition	displacement at any point is the vector sum of the individua
Principle of	The principle of superposition states that the resultant
Radiation	and magnetic components.  Chapter 11: Superposition
Electromagnetic	A self-propagating transverse wave in space with electric
Polarised Waves	All particles vibrate in the same plane at all times
	wave.
	move in a direction parallel to the direction of travel of the
Longitudinal Waves	A longitudinal wave is one in which particles of the medium
	of the wave.
	move in a direction perpendicular to the direction of trave
Transverse Waves	A transverse wave is one in which particles of the mediun
	particles in the same wave or another wave.
Phase	The stage of motion of the particle with respect to other
Intensity, I	The amount of energy incident per unit area per unit time.
	troughs.
	waveforms, such as two successive crests or two successive
Wavelength, λ	The distance between corresponding points in successive
	Chapter 10: Wave Motion
	SECTION IV: WAVES
	intermolecular attractive or repulsive forces.
	molecules are perfectly elastic and which there are n
deal Gas	A gas in which all collisions between the atoms ar
, out supress,	the body by 1K.
Heat Capacity	The <u>quantity of heat</u> required to <u>raise the temperature</u>
riple Foliit of Water	states of water can co-exist in equilibrium.
riple Point of Water	The particular temperature and pressure at which the three
	system and the work done on the system.
	system is equal to the sum of the heat supplied to the

Electric Field Strength, E	The electric field strength at a point is defined as the force
	per unit charge acting on a small positive test charge placed
	at that point.
Coulomb's Law	The force between two point charges is directly
	proportional to the product of the charges and inversely
	proportional to the square of the distance between the
	charges.
Uniform Electric Field	Electric field strength is equal in magnitude and has the
	same direction at all points in the region.
Electric Potential, φ	The electric potential at a point in an electric field is the
	work done per unit charge in bringing a positive test charge
	from infinity to the point (without a change in kinetic
	energy).
	Chapter 13: Current of Electricity
Electric Current, I	The net amount of charge passing through a point per unit
	time.
Coulomb, C	One coulomb is the <u>quantity of electric charge</u> that passes a
	given point in a circuit in one second when there is a
	constant current of one ampere.
Potential Difference, E	The potential difference between two points in a circuit is
	the amount of electric energy that is converted to other
	forms of energy when a unit charge passes from one point
V-16 V	to the other.
Volt, V	One volt is the potential difference between two points in a
	circuit in which one joule of energy is converted when one
Desistance D	coulomb of charge passes from one point to the other.  The electrical resistance of a conductor is defined as the
Resistance, R	ratio of the p.d. across it to the current through it.
Ohm O	One ohm is defined as the resistance of a conductor in
Ohm, Ω	which a current of one ampere passes through it when the
	p.d. across it is one volt.
Electromotive Force	The e.m.f. of a source is defined as the amount of energy
(e.m.f.)	converted from other forms to electrical energy when the
(,	source drives a unit charge round a complete circuit.
Ampere, A	One coulomb per second.
	Chapter 15: Electromagnetism
Magnetic Flux Density, B	The flux density of a magnetic field is the force per unit
	length on a straight conductor carrying unit current placed
	perpendicularly to the field. (RJCCT208)
Tesla, T	The magnetic flux density of a magnetic field is one tesla if
	the force acting on 1m length of a conductor carrying 1A of
	<u>current</u> placed <u>perpendicular</u> to the field is <u>1N</u> .
Electronvolt, eV	The electronvolt is the energy gained by an electron when it
	is accelerated through a p.d. of one volt.
C	hapter 16: Electromagnetic Induction
Magnetic Flux, ø	Magnetic flux through a plane surface is the product of the

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3	area and the magnetic flux density that passes through the
	area perpendicularly.
Weber	One weber is the magnetic flux if a field of flux density one
	tesla exists at right angles to an area of one metre square.
Magnetic Flux Linkage, Φ	The magnetic flux linkage of a coil is the magnetic flux
	passing through each turn of the coil multiplied by the
	number of turns of the coil.
Faraday's Law	Whenever there is a change in magnetic flux linkage of a
	circuit or coil, an e.m.f. is induced in the circuit and the
	magnitude is directly proportional to the rate of change of
	magnetic flux linkage of the circuit or coil.
Lenz's Law	Lenz's Law states that the direction of the induced current is
	such as to oppose the change in flux which causes it.
	Chapter 17: Alternating Current
RMS value of an	It is the value of the steady direct current which would
alternating current	dissipate heat at the same rate in a given resistance as the
Mara Da	alternating current.
Mean Power	The mean power dissipated by a resistive load is half the
Cinucaidal AC/Valence	maximum power for a sinusoidal AC.
Sinusoidal AC/Voltage	Current/voltage varies periodically with time in magnitude
Rectification of an AC	and direction.
Half-wave Rectification	Conversion of AC to DC.
nan-wave Rectification	For half the cycle, the diode allows current to flow but for the other half of the cycle, the current flow is very small due
*	to the high resistance of the diode being in reverse bias.
	SECTION VI: MODERN PHYSICS
	Chapter 18: Quantum Physics
Photoelectric Effect	The emission of electrons from a metal as a result of light
	with sufficiently short wavelength falling on it.
Work function, Φ	The work function of a material is defined as the minimum
	amount of energy required to remove a free electron from
	the surface of a material
Square of the absolute	Probability density of finding the particle at a particula
magnitude of the Wave	point, at a particular time.
Function, $ \Psi ^2$	
Potential Barrier	A potential barrier is a region within which the potentia
	energy of the particle is much higher than immediately
	outside it.
Photon	A guantum of electromagnetic energy.
Ionisation Energy	The ionization energy of an atom is the minimum energy
	required to remove an electron completely from the atom.
Ionisation	The process of creating charged particles.
Transmission Coefficient	The probability of the particle being transmitted.
Reflection Coefficient	The probability of the particle being reflected. hapter 19: Lasers and Semiconductors
	ABROR IN LACORE AND EDMICONDUITE AFF

	discasi vil
Stimulated Emission	direction without any external stimulation.
Stimulated Emission	An incoming photon, whose energy is exactly equal to the
	excitation energy of the atom, induces the excited atom to
	fall to a lower energy level and releases a photon in the
	process. This photon released is similar to the one which
	induces its emission. The two photons are emitted at the
	same time and in the <u>same direction</u> .
Population Inversion	When there are more atoms in the excited state than in the
	ground state.
Intrinsic Semiconductor	A semiconductor without added impurities.
<b>Extrinsic Semiconductor</b>	A semiconductor with added impurities.
P-N Junction	A P-N Junction is a single semiconductor crystal that has
	been selectively doped so that one region is n-type material
	and the adjacent region is p-type material.
Stimulated Absorption	When an atom at a lower energy level absorbs a photon and
	moves to a higher energy level.
	Chapter 20: Nuclear Physics
Nucleon Number (Mass	The number of nucleons (protons and neutrons) in the
Number)	nucleus.
Proton Number (Atomic	The number of protons in the nucleus.
Number)	
Mass Defect	The difference between the sum of the individual masses of
	protons and neutrons and the mass of a nucleus.
Binding Energy	The amount of energy needed to split a nucleus into its
	individual nucleons.
Binding Energy per	Binding energy divided by the mass or nucleon number of
Nucleon	the nucleus.
Nuclear Fusion	Process by which nuclei with mass numbers lower than 56
	combine to form nuclei of higher mass numbers which are
	more stable.
Nuclear Fission	Process by which nuclei of mass numbers larger than 56
	break up to form lighter nuclei which are more stable.
Activity, A	The <u>number of atoms</u> of a radioactive substance that <u>decay</u>
300	per unit time.
Decay constant	The probability of decay per nucleus per unit time.
Half life	Half life of a radioactive element is the time taken for a
	sample of atoms to decay to half their initial number.
Neutron Number	The number of <u>neutrons</u> in the nucleus.
Radioactivity	The spontaneous emission of $\alpha$ , $\beta$ or $\gamma$ radiation by a parent
*	nucleus which results in itself being transformed into a
	completely different <u>daughter nucleus</u> .